Ultrasonic dyeing of wool fabric with aqueous extract of Ratanjot (*Onosma echioides*) natural dye

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Abstract
The purpose of this study is to compare the conventional and ultrasonic dyeing methods in terms of its colour strength and colour fastness properties of wool fabric dyed with Ratanjot root. The results suggested that dyeing of wool fabric with natural dye Ratanjot (*Onosma echioides*) using ultrasonic waves, significantly improved the dye uptake percentage to 12.31 per cent from conventional heating methods. The fastness grade were found to be higher with ultrasonic than conventional heating. Additionally, the fabric dyed using ultrasonic waves gave a deeper shade and good colour intensity even at lower dyeing time (75min) and temperature (60°C). Therefore, ultrasound wave represents a promising technique for increasing diffusion of dye by the effect of cavitation, as well as for improving the effectiveness of processes when compared to conventional heating.

Key words: Conventional, Colour Strength, Colour fastness, Ratanjot, Ultrasonic

Introduction
Natural dyes are colourants obtained from natural resources such as plants, minerals and animals. The dyeing of natural colourants was considered as one of oldest techniques, practiced by ancient people before thousands of years (Prabhu and Bhute, 2012). The ancient arts namely Ajanta, Ellora, Sithannavasal, Mithila wall paintings (mural art) and Egyptian pyramids were done with natural colourants (Mukherjee and Kanakarajan, 2017). Besides, Vedas also cited red, yellow, blue, black and white as main dyeing colours and stated that, the ancient craftsmen dyed blue from indigo, yellow from turmeric and saffron, brown from cutch and red from lac, safflower and madder. But later, synthetic dye was introduced in the dye market considerably due to its cheaper cost, wide range of colours and good fastness properties (Paisan *et al*., 2002; Saravanan and Chandramohan, 2011). However, during the last few decades, the use of synthetic dyes is gradually decreasing due to increase environmental awareness and harmful effects because of either toxicity or their non-biodegradable nature. In addition to above, some serious health hazards like allergy and carcinogenicity are associated with the synthetic dyes. As a result, recently a ban has been enforced all over the world including European Economic Community (EEC), Germany, USA and India on the use of some synthetic dyes (Yusuf *et al*., 2017). Therefore, natural dye started regaining interest in textile industries for its biodegradable, non-toxic and environment friendly nature (Yusuf *et al*., 2016). The name Ratanjot (*Onosma echioides*) is attributed to the roots of various Boraginaceaous plant species. Traditionally, it is used as a food colourant in cosmetic formulations and pharmaceutical preparations. Several naphthoquinone pigments are present in the bark of ratanjot roots, which give a violet red colour. The main pigment present in ratanjot root is alkanin (Fig.1). It is regarded as one of the important herbal drugs of indigenous systems of medicine. The root bark of ratanjot is reported to have several biological properties such as antibacterial, antifungal, antiviral, antioxidant, radical scavenging and antithrombotic activity (Arora *et al*., 2009). Previously, many researchers reported ratanjot as one of the potential natural dyes sources (Gupta and Saini, 2018; Chattopadhyaya *et al*., 2018; Chattopadhyaya *et al*., 2013). Traditionally, conventional boiling method is used for the dyeing textiles with natural dyes; however, there are certain limitations such as poor colour strength,
complexity of dyeing process, longer time for dye extraction, poor colour fastness and reproducibility of shades. Thus, it is necessary to explore a suitable alternative technique for more efficient natural dyeing to improve the dyeing properties of natural dyes (Rahman et al., 2013). In this study, ultrasonic cleaner approach was used as suitable method for dyeing. Ultrasound dyeing method saves energy by dyeing at lower temperature or shortens the process, lowers the consumption of auxiliary chemical, lesser processing costs, which lead to increase in the competitiveness in the industry, process improvement by controlling the colour shade (Parvinzadeh et al., 2009). The ultrasonic dyeing mechanism involves formation of cavitation, compression or rarefaction and streaming as shown in Fig.2. Due to cavitation (the phenomenon of collapse and expansion of micro bubbles in the liquid medium), ultrasound improves dyeing process and cavitation process give enhancement in four ways - (i) breaking-up aggregates of dye in solution which would increase their rate of diffusion into the fibre, (ii) removing air from between fibres to improve fibre–liquid contact, (iii) disrupting the fibre–liquid boundary layer to increase the rate of dye diffusion into the fibre and (iv) increasing the swelling of fibres which would increase the rate of dye diffusion through the fibre and increase in the area of the fibre–liquid interface. Besides, the pollution parameters such as BOD (Biochemical Oxygen Demand), COD (Chemical Oxygen Demand) and TOC (Total Organic Carbon) levels in textile dye house effluents are found to be decreased after using ultrasonic dyeing method. The dyeing processes on cotton, wool, silk, acrylic, nylon and polyester with the use of ultrasound energy have dye diffusion coefficient of 30 per cent and reduces the dyeing time by 20 per cent (Udrescu et al., 2014). A lot of work has been carried out for dyeing of textiles with natural dye using conventional heating but the work on ultrasonic technique is very limited. One study reported that ultrasound approach improved dyeability as well as colour fastness properties of dyed fabric (Kamel et al., 2011). Similarly another study also reported that the use of ultrasound wave improved the dye uptake with lower temperature, i.e, 60°C (Mansour and Haffernan, 2011). Therefore, in the present study, wool fabric was dyed with aqueous extract of ratanjot dye using both conventional and ultrasonic dyeing methods and compared it in terms of its colour strength, CIE Lab values and colour fastness properties.

Figure 1: Alkanin: the main colour component.

Figure 2: Schematic diagram of ultrasonic dyeing mechanism: (a) Cavitations or bubbles formed in liquid by ultrasonic waves (b) Forces created during compression or rarefaction (c) Rapid movement of liquids by ultrasonic pressure
Materials and Methods
Preparation of samples: A plain weave wool fabric was procured from the local market of Ludhiana, Punjab. Roots of Ratanjot plant were collected from Herbal garden of Punjab Agricultural University campus. The root parts were dried under the sun and finely ground into powder form. Four types of mordants namely Amla, babool gum powder (Natural mordants) and Alum and tannic acid (Synthetic mordants) were used. Analytical grade urea, sodium chloride, sodium carbonate and acetic acid were used in the experiments.

Optimisation of dyeing parameters: Experiments were conducted for optimizing ultrasonic dyeing and mordanting conditions of wool with roots of ratanjot dye using four different natural and synthetic mordants. The conditions namely - dye extraction time (15, 30, 45, 60 minutes), dye extraction temperature (30, 45 and 60 °C), dyeing concentration (1, 2, 3, 4 and 5g/g), dyeing time (30, 45, 60, 75 and 90 minutes), dyeing pH (4-8) and dyeing temperature (30, 45 and 60 °C) were optimized. Mordant concentration of 0.1, 0.2, 0.3, 0.4 and 0.5gm of amla and babool mordants were mordanted simultaneously at optimum dyeing pH, for optimum time at optimum temperature. The concentrations of alum were 5, 10, 15, 20 and 25gm/100g of fabric. The concentrations of tannic acid were 1, 2, 3, 4 and 5 per 100g of fabric.

Final dyeing of wool fabric by using Conventional and Ultrasonic dyeing methods: Conventional dye extraction was carried out as described by Bains et al. (2002). The optimized dyeing conditions for conventional dyeing technique were dye concentration of 3g/g of fabric, dyeing pH 7, dyeing time of 90 min and 100°C dyeing temperature. The optimum mordant concentration of of amla, babool, alum and tannic acid were 0.5, 0.2, 15 and 4g/g of fabric respectively. The dye solution was poured in the ultrasonic cleaner bath, keeping the Material to liquor ratio (MLR) of 1:250. A pre-soaked wool fabric was immersed into the dye solution and subjected to ultrasound at optimized temperature for optimized time. The simultaneous mordanting method was performed at optimized mordant concentration. After dyeing, the baths were allowed to cool down for some time to avoid sudden change during rinsing. The samples were taken out, washed in mild detergent solution and rinsed in tap water to remove unfixed dye material. After thoroughly rinsing, the samples were finally dried in shade. The dyeing with ultrasound was carried out at 220V/50Hz.

Evaluation of colour strength: The colour strength (K/S) of dyed fabric is a measure of dye concentration on the fabrics. The colour strength expressed as the K/S value in the wavelength range 400–700 nm at 10 nm intervals within the visible spectrum was calculated by the following Kulbelka – Munk equation:

\[
K/S = \frac{(1-R)^2}{2R}
\]

Where “R” is the reflectance at the wavelength of maximum absorption, “K” is absorption coefficient and “S” is the scattering coefficient. The higher the K/S value, the greater is the dye uptake, resulting in better colour yield.

Evaluation of CIE Lab value: CIE Lab (L*, a*, b*) values of the dyed samples were recorded with illuminant D65/10° observer using Colorflex Hunter Lab. The measurement was replicated four times for each sample. Colour coordinates CIE L*a*b* were determined to define the properties of color on dyed fabric. CIE L*a* b* values were used to evaluate the lightness, red/green, and blue/yellow characteristics of the dyed samples. L* is corresponding to the brightness (100 = white, 0 = black), a* is corresponding to the red/green coordinate (positive sign= red, negative sign = green), and b* is corresponding to the yellow/blue coordinate (positive sign= yellow, negative sign = blue). From the L*, a*, and b* coordinates, chroma (C*) values were calculated by using the following equation:

\[
C* = \sqrt{a^*^2 + b^*^2}
\]

Chroma measures the intensity or saturation of the colourant.

Evaluation of Colour fastness: The light, wash, rubbing and perspiration fastness values of all dyed samples were determined by Digital light fastness tester, Launderometer, Crockmeter and Perspirometer, tested according to ISO test methods.

Statistical analysis: The optical density and K/S values of each dyeing condition were evaluated by analysis of mean and each treatment was replicated
three times. The standard error of the difference SED (±) was calculated for each mordant.

Results and Discussion
Various experiments were carried out to compare the conventional and ultrasonic methods for dyeing of wool fabric in terms of dye uptake and colour fastness properties.

Optimized dyeing conditions: Fig.3. indicates that the value of optical density gradually increased with increase in extraction time of dye. It was also observed that the optical density was higher at dye extraction time of 60 minutes. Therefore, 60 minutes is taken as the optimum dye extraction time.

The results obtained from Figure 4 revealed that optical density of dye solution increased with increase in extraction time. It was observed that the optical density was maximum at 60°C. From the Fig.5, it is clear that dye absorption increased when the dye material was increased from 1 to 2 g/g of wool fabric, but decrease in dye absorption with increase in the dye concentration from 3 to 5g / 100 g of wool. Therefore, 2 g/g was the optimum dye concentration of wool. K/S values for different dyeing pH furnished in Fig.6 revealed that the dye absorption increased from dyeing pH from 4 to 5. But there was gradual decrease in the dye absorption when dyeing pH increased from 6 to 8. The maximum K/S value was found at dyeing pH of 5. It was also observed that the dye absorption increased as the dyeing temperature increased as shown in Fig.7. Dye absorption was maximum at 60°C and considered as optimum dyeing temperature. It could be envisaged from the Fig.8 that initially the values of K/S increased when dyeing time was increased from 30 to 75 minutes. The K/S values were decreased when the dyeing time increase from 75 to 90 minutes. The colour strength was found to be maximum at dyeing time of 75 minutes. Therefore, the maximum dyeability was found at dye extraction time 60 minutes and dye extraction temperature 60°C. And, the maximum dye was absorbed at dye material concentration 2g/g of wool, dyeing pH 5, dyeing temperature 60°C and dyeing time 75 minutes.

Interestingly, when compared, the optimized dyeing conditions of ultrasonic and conventional dyeing methods, it was found that there was reduction of dye concentration requirement, dyeing time and dyeing temperature by using ultrasonic waves since the optimized dye concentration, dyeing time and dyeing temperature of conventional dyeing method on wool were 3g/g of fabric, 90 minutes and 100°C respectively. Therefore, ultrasonic dyeing method saves energy, time and amount of dye requirement.

The optimum mordant concentration values have been presented in table 1. The optimum value for amla was found to be 0.4 g/g, whereas, for babool, it was 0.5g/g. For alum it was 15g/g and 4 g/g for tannic acid. Tannic acid mordanted samples showed maximum K/S values followed by amla, babool and alum.

Evaluation of Colour strength: The colour strength (K/S values) of wool fabric using both conventional and ultrasonic dyeing methods was shown in Fig. 9. Comparing the conventional and ultrasonic methods, the colour strength value obtained by ultrasonic dyeing method was higher than with conventional method. This was due to the formation of microscopic bubbles, or cavitations by the ultrasonic waves in the liquid system. When these bubbles are collapsed, they generate tiny but powerful shock waves. The release of micro bubbles and shock waves led to promote the action and binding between the dye particles and wool fibre. Likewise, one study also observed that the ultrasonic dyed samples with Drimerene Blue CL-BR and Drimarene Red Cl-5B improved its colour strength than the conventional dyed samples (Syed et al., 2013). It was also observed that K/S value of unmordanted sample was more than the sample mordanted with the alum on wool. Mansour and Heffernan (2011) also found that the value of colour strength of silk fabric dyed with sticta coronata lichen declined when mordanted with 10g of alum in the presence of ultrasound. The reason may be due to formation of cluster/ aggregation of alum solution that rupture the interaction of structure thus hinders formation of bond between the metal ions and dye molecules under the ultrasonic energy. Maximum colour strength was found to be in sample mordanted with tannic acid on wool.

Evaluation of CIE Lab Values: CIE Lab values for dyed fabrics with Ratanjot in both conventional and ultrasonic methods with different mordants were given in Table 2. The results revealed that the L* value of all the ultrasonic dyed samples were
Figure 3: Optimization of dye extraction time of Ratanjot dye

Figure 4: Optimization of dye extraction temperature of Ratanjot dye

Figure 5: Optimization of dye concentrations of Ratanjot dye

Figure 6: Optimization of dyeing pH of Ratanjot dye

Figure 7: Optimization of dyeing temperature of Ratanjot dye

Figure 8: Optimization of dyeing time of Ratanjot dye

Figure 9: Comparison of the K/S value of dyed wool fabric using both conventional and ultrasonic dyeing method

CD = Conventional dyed sample, UD = Ultrasonic dyed sample

The higher and positive a* value of the all the ultrasonic dyed samples showed that the samples were redder as compared to conventional dyed samples. The ultrasonic samples mordanted with amla were redder, followed by babool, unmordanted, alum and tannic acid samples. In
Table 1: Optimization of mordant concentrations for Ratanjot dye on wool

<table>
<thead>
<tr>
<th>Mordant concentration g/100g wool of Amla</th>
<th>K/S value</th>
<th>Mordant concentration g/100g wool of Babool</th>
<th>K/S value</th>
<th>Mordant concentration g/100g wool of Alum</th>
<th>K/S value</th>
<th>Mordant concentration g/100g wool of Tannic acid</th>
<th>K/S value</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.1</td>
<td>6.03</td>
<td>0.1</td>
<td>4.99</td>
<td>5</td>
<td>3.77</td>
<td>1.0</td>
<td>8.17</td>
</tr>
<tr>
<td>0.2</td>
<td>6.41</td>
<td>0.2</td>
<td>5.02</td>
<td>10</td>
<td>4.01</td>
<td>2.0</td>
<td>8.70</td>
</tr>
<tr>
<td>0.3</td>
<td>6.80</td>
<td>0.3</td>
<td>5.86</td>
<td>15</td>
<td>4.12</td>
<td>3.0</td>
<td>9.26</td>
</tr>
<tr>
<td>0.4</td>
<td>7.22</td>
<td>0.4</td>
<td>6.40</td>
<td>20</td>
<td>3.89</td>
<td>4.0</td>
<td>10.53</td>
</tr>
<tr>
<td>0.5</td>
<td>6.80</td>
<td>0.5</td>
<td>6.60</td>
<td>25</td>
<td>3.56</td>
<td>5.0</td>
<td>9.87</td>
</tr>
</tbody>
</table>

Table 2: Chromaticity values of the dyed wool fabric

<table>
<thead>
<tr>
<th>Samples</th>
<th>L*</th>
<th>a*</th>
<th>b*</th>
<th>C*</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>CD</td>
<td>UD</td>
<td>CD</td>
<td>UD</td>
</tr>
<tr>
<td>Unmordanted</td>
<td>69.63</td>
<td>68.22</td>
<td>3.44</td>
<td>7.19</td>
</tr>
<tr>
<td>Amla</td>
<td>73.59</td>
<td>71.58</td>
<td>8.33</td>
<td>9.82</td>
</tr>
<tr>
<td>Babool</td>
<td>52.14</td>
<td>43.56</td>
<td>7.78</td>
<td>9.35</td>
</tr>
<tr>
<td>Alum</td>
<td>57.90</td>
<td>51.00</td>
<td>2.69</td>
<td>4.33</td>
</tr>
<tr>
<td>Tannic Acid</td>
<td>46.64</td>
<td>41.98</td>
<td>1.35</td>
<td>1.79</td>
</tr>
</tbody>
</table>

CD= Conventional Dyed sample, UD= Ultrasonic Dyed sample, L*= Lighter/ darker, a*= Redder/ greener, b*= Yellower/ bluer, C*= Intensity/ saturation of the colorant

Table 3: Colour fastness grades of conventional and ultrasonic wool dyed using Ratanjot dye

<table>
<thead>
<tr>
<th>Samples</th>
<th>Light fastness grades</th>
<th>Washing fastness grades</th>
<th>Rubbing fastness grades</th>
<th>Perspiration fastness grades</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>CC</td>
<td>CS</td>
<td>Dry</td>
<td>Wet</td>
</tr>
<tr>
<td>Unmordanted</td>
<td>CC</td>
<td>CC</td>
<td>CC</td>
<td>CC</td>
</tr>
<tr>
<td>CDS</td>
<td>2</td>
<td>4/5</td>
<td>5</td>
<td>4/5</td>
</tr>
<tr>
<td>UDS</td>
<td>2/3</td>
<td>4/5</td>
<td>5</td>
<td>5</td>
</tr>
<tr>
<td>Amla</td>
<td>2</td>
<td>5</td>
<td>5</td>
<td>5</td>
</tr>
<tr>
<td>UDS</td>
<td>2</td>
<td>5</td>
<td>5</td>
<td>5</td>
</tr>
<tr>
<td>Babool</td>
<td>2/3</td>
<td>4/5</td>
<td>5</td>
<td>5</td>
</tr>
<tr>
<td>UDS</td>
<td>3</td>
<td>5</td>
<td>5</td>
<td>5</td>
</tr>
<tr>
<td>Alum</td>
<td>2</td>
<td>3</td>
<td>4</td>
<td>5</td>
</tr>
<tr>
<td>UDS</td>
<td>2</td>
<td>4</td>
<td>4/5</td>
<td>5</td>
</tr>
<tr>
<td>Tannic acid</td>
<td>2</td>
<td>3</td>
<td>4</td>
<td>5</td>
</tr>
<tr>
<td>UDS</td>
<td>2/3</td>
<td>4</td>
<td>4/5</td>
<td>5</td>
</tr>
</tbody>
</table>

C = Cotton, W= Wool, CC = Color Change, CS = Color Staining, CDS = Conventional Dyed Sample, UDS = Ultrasonic Dyed Sample

Both the techniques, the a* value of samples mordanted with alum and tannic acid showed light red as compared to unmordanted (control) sample. It is clear from the table 2, that the higher b* value of ultrasonic dyed samples mordanted with amla and tannic acid samples showed more yellowish as compared to their conventional dyed sample whereas, lower b* values of ultrasonic dyed unmordanted, mordanted samples with babool and alum indicated that they have light yellow colour as compared their conventional dyed samples. The C* values of conventional dyed samples were higher.
than the samples obtained by ultrasonic waves except the amla and tannic acid mordanted samples, which indicates that the samples obtained by ultrasonic waves are more saturated as compared to conventional boiling.

**Evaluation of colour fastness properties:** The light, washing, rubbing and perspiration fastness grades of wool samples dyed with Ratanjot dye have been presented in the table 3.
The light fastness properties of conventional dyed samples was 2, indicated poor and improved to fair when the sample dyed by using ultrasonic dyeing technique. The washing fastness grades in terms of colour change were improved because of ultrasonic wave. No staining was observed on wool for both conventional and ultrasonic dyed sample except for unmordanted sample. Therefore, ultrasonic dyed samples have better washing fastness properties as compared to conventional dyed samples. In ultrasonic dyeing techniques, dye molecules are not fixed on the fibre surface but it reacted with the fibre surface. Similarly, the rubbing fastness and perspiration grade of conventional dyed was improved in ultrasonic dye samples. Another study also found that rubbing fastness properties of cotton, silk and wool fabric give the better result when dyed with *Malus sikkimensis* by using ultrasonic dyeing techniques as compared to conventional dyeing techniques (Vankar et al., 2009).

**Conclusion**
In this study, the dye extraction and dyeing conditions of wool fabric with ratanjot natural dye were optimised. Based on the optical density and K/S values, the optimum dye extraction conditions of ultrasonic dyeing methods were: extraction time 60 min and extraction temperature 60°C while the optimized dyeing parameters were: dyeing temperature 60°C, dye concentration 2g/g, dyeing time 75 min and dyeing pH 5. Interestingly, the use of ultrasonic wave in the dyeing process was found to be significant improvement in the colour strength (K/S) and colour fastness properties as compared to conventional heating method. In addition to advantage of dye absorption and colour fastness, fabrics dyed by using ultrasonic dyeing method showed a deeper shade colour and good colour intensity than conventional dyeing method. Moreover, the textile dyeing industry has long been struggling to cope with high energy costs, rapid technological changes and the need for a faster delivery time. The effective management of ultrasonic technology can reduce the energy costs and improve productivity. Though the initial capital cost of installing ultrasonic dyeing machines would be high but it will ultimately save energy by dyeing at low costs. In conventional dyeing, high dyeing temperatures are maintained for which lot of energy is used. Moreover, many synthetic mordants and auxillaries are required in conventional dyeing which not only adds to the cost but also pollutes the environment. In ultrasonic dyeing, there will be reduced consumption of auxiliary chemicals which will lower the processing costs and also lessen the processing time. Hence, it will be a win-win situation for both the manufacturer and the environment.

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**References**


