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Agronomic biofortification with zinc and iron to enhance nutrient concentrations in mango

Archana Mahida 🖂

Department of Fruit Science, ASPEE College of Horticulture and Forestry, Navsari Agricultural University, Navsai, India Tandel, Y. N

Department of Fruit Science, ASPEE College of Horticulture and Forestry, Navsari Agricultural University, Navsai, India Patel, D. P

Department of Fruit Science, ASPEE College of Horticulture and Forestry, Navsari Agricultural University, Navsai, India **Manish Kumar**

Department of Horticulture, School of Life Sciences, Central University of Tamil Nadu, Thiruvarur, India

ARTICLE INFO	ABSTRACT
Received : 18 January 2023	Biofortification is a global-scale agricultural approach that can improve
Revised : 18 June 2023	human nutrition. Agronomic biofortification is viewed as a quick-fix and
Accepted : 04 July 2023	supplemental approach. Agronomic biofortification, especially foliar
	application, is highly effective for zinc and iron. A field experiment on
Available online: 12 November 2023	agronomic biofortification of zinc and iron micronutrients in mango cv. Kesar
	was carried out in 2016-2017 at the Regional Horticultural Research Station,
Key Words:	ASPEE College of Horticulture and Forestry, Navsari Agricultural University,
Agronomical fortification	Navsari (Gujarat). The experiment was arranged in a completely randomized
ZnSO ₄	design (CRD) with three replications containing 9 treatments. The results show
FeSO ₄	that foliar application of ZnSO4 and FeSO4 (0.5% each) resulted in higher N
foliar spray	(48.73 mg/100 g) and K (94.17 mg/100 g) in the pulp and P (0.056%) in the peel
N, P and K	of mango. The iron (Fe) and zinc (Zn) contents in pulp and peel were highest in
	treatment T ₉ (0.50% FeSO ₄ + 0.50% ZnSO ₄), which was on par with those in
	treatment T ₈ (0.50% FeSO ₄ + 0.25% ZnSO ₄).

Introduction

Mango (Mangifera indica L.) belongs to the family years to come (Cakmak, 2008). Iron is a crucial Anacardiaceae. It is a significant fruit crop in Asia and has gained recognition on a global scale. It has long been associated with culture and religion as a helpful and delectable fruit. Mango is known as "The King of Fruits" due to its great nutritional content. Micronutrients are crucial for plant metabolism and have negative consequences when they are lacking, which makes them important for development. Micronutrients crop have а significant impact on plant development in addition to disease resistance in farmed crop species. Due to widespread prevalence of micronutrient the imbalance throughout India, the situation has taken a troublesome turn. Due to the significance of micronutrients in the diet of humans, it is anticipated that biofortification of horticulture crops will play an important and decisive role in addressing the nation's nutritional security in the

mineral and part of several proteins involved in metabolism and oxygen transport. The body typically regulates the amount of iron absorbed from the diet to keep its levels of iron regular. The greatest global nutritional problem, according to the WHO, is iron deficiency. Iron deficiency is responsible for approximately 50% of anemia worldwide. Globally, more than 2 billion people suffer from iron deficiency (Mehansho, 2006). In addition to being a vital component of many enzymes, zinc also plays a crucial role in cellular development and differentiation in tissues with a high rate of differentiation and turnover, such as the immune system and the digestive tract. The skin, brain, central nervous system, and immunological, skeletal, and reproductive systems are all impacted by zinc deficiency. Over 2 billion individuals, according to the World Health Organization

Corresponding author E-mail: mahidaarchana24@gmail.com Doi: https://doi.org/10.36953/ECJ.17432548

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(WHO), are believed to be zinc deficient. Approximately 20% of the world's population may be at danger of zinc insufficiency, according to estimates of zinc consumption and bioavailability obtained from FAO food balance statistics. Iron and zinc deficits were listed as two of the world's most significant health risks in the World Health Report from 2000.Increasing the content of vitamins in fruit while concurrently enhancing their bioavailability is the dual goal of biofortification, which aims to reduce human micronutrient deficiencies. It is a possible sustainable and costeffective agronomic method to increase the micronutrient content of food. A future plan is to micronutrient deficiencies globally. address Conferring to estimates from the World Health Organization (Anon., 2000), biofortification of iron might help treat two billion individuals with anemia caused by iron deficiency. Maintaining enough zinc and iron delivery to the fruit throughout the reproductive stage and maintaining a sufficient amount of readily accessible zinc and iron in foliar solution appear to be key agronomic fortification strategies. It is a highly appealing and practical technique for efficiently addressing global health issues connected to iron and zinc deficiencies.

Material and Methods

Study Area: The experiment was carried out on a 20-year-old mango orchard planted at a 10 m×5 m distance, and the site was located at the Regional Horticultural Research Station, ASPEE College of Horticulture and Forestry, Navsari Agricultural University, Navsari (Gujarat) from October 2016 to June 2017. The experimental site was situated at 20°57' north latitude, 72° 54' east longitude and has an altitude of 10 meters above the mean sea level. The analytical procedures employed for determining nutrients are presented in Table 1. The climate of this area is characterized by three well-defined seasons: monsoon, mild winter and summer. The monsoon commences from the second week of June and lasts up to the end of September. The intensity of rainfall is high during July and August. The details of the meteorological data with respect to maximum and minimum temperature, relative humidity, rainfall, etc., during the period of experimentation were obtained from the Agricultural Meteorological

Observatory, College Farm, N. M College of Agriculture, NAU, Navsari.

Treatment details: The trial was directed in a completely randomized design (C.R.D.) with 9 treatments and replicated three times. The details of the experimental treatment plan employed in the present investigation are as follows: Control (T_1) , 0.25% FeSO₄ (T₂), 0.50% FeSO₄ (T₃), 0.25% ZnSO₄ (T₄), 0.50% ZnSO₄ (T₅), 0.25% FeSO₄ + 0.25% ZnSO₄ (T₆), 0.25% FeSO₄ + 0.50% ZnSO₄ (T₇), 0.50% FeSO₄ + 0.25% ZnSO₄ (T₈), and 0.50% $FeSO_4 + 0.50\%$ ZnSO₄ (T₉) (Table 2). The foliar spray of these nutrients was performed at the time of flowering (21/12/2016), pea stage (17/02/2017) and egg stage (20/03/2017). For the preparation of Zn and Fe foliar solutions, commercial grade ZnSO₄.7H₂O and FeSO₄.7H₂O fertilizers were used, respectively. The mandatory quantities of ferrous sulfate and zinc sulfate were weighed and dissolved in water, and then the pH of the solution was adjusted to 6.0 by using saturated CaCO₃ solution.

Table 1:- Analytical procedures employed fordetermining nutrients from pulp and peel of mango

Nutrient	Method employed	Reference
N	Wet digestion (Chromic acid)	Trivedi et
		al. (1999)
P and K	Wet digestion (diacid) followed	Jackson
	by P: Spectrophotometric	(1973)
	(Vanadomolybdophosphoric	
	yellow color method) K: Flame	
	photometric	
Fe, Mn,	Atomic Absorption	Elwell and
Zn and Cu	Spectrophotometer (AAS)	Gridley
		(1967)

 Table 2: Micronutrient fertilizers and water required for foliar spraying

S	Treatments	Fertilizers required (g/20 Lit)	
Ν	1 reatments	FeSO ₄ .7H ₂ O	ZnSO ₄ .7H ₂ O
1.	Control	-	-
2.	0.25% FeSO4	91.4	-
3.	0.50% FeSO4	182.8	-
4.	0.25% ZnSO4	-	89
5.	0.50% ZnSO4	-	178
6.	0.25% FeSO ₄ +		
	0.25% ZnSO4	91.4	89
7.	0.25% FeSO ₄ +		
	0.50% ZnSO4	91.4	178
8.	0.50% FeSO ₄ +		
	0.25% ZnSO4	182.8	89
9.	0.50% FeSO ₄ +		
	0.50% ZnSO4	182.8	178

193 Environment Conservation Journal

Statistical analysis

The acquired data were statistically evaluated in accordance with the method (Panse and Sukhatme, 1967) suitable for a completely randomized design, and the treatment means were compared using critical differences at a 5% level of probability.

Results and Discussion

The N and K contents in the mango pulp were 48.73 mg/100 g and 94.17 mg/100 g, respectively, which were highest in the combined application of 0.50% FeSO4 + 0.50% ZnSO4 (T9) compared to the other treatments and controls. However, the P content did not show any significant response to foliar spraying on the Fe and Zn contents in the

pulp of mango. Furthermore, the N and K contents in the peel of mango did not demonstrate significant variation by the foliar spray treatments applied. However, a significantly increased P content in the peel (0.056 ppm) was found in the foliar spray of 0.50% FeSO₄ and 0.50% ZnSO₄. Increased nutrient content in pulp and peel may be the result of foliar micronutrient spray, which reduced nutritional deficits and enhanced fruit quality due to absorption of macroand micronutrients in tissues and organs of mango plants. Singh et al. (2020) found a similar effect with mango. The maximum zinc (Zn) content and iron (Fe) content in the pulp and peel of mango cv.

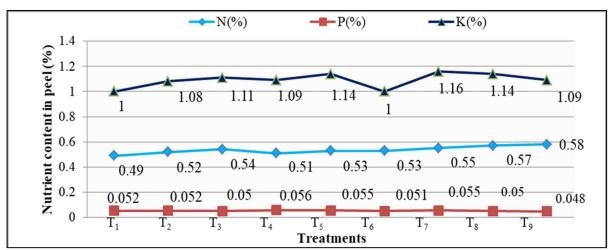


Figure 1: Effect of foliar spray of Zn and Fe on major nutrient content in peel of mango cv. Kesar

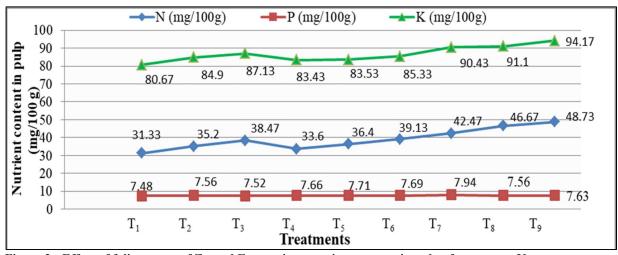


Figure 2: Effect of foliar spray of Zn and Fe on micro nutrient content in pulp of mango cv. Kesar

¹⁹⁴ Environment Conservation Journal

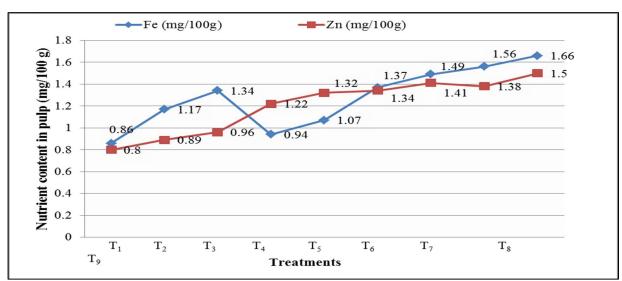


Figure 3: Effect of foliar spray of Zn and Fe on micro nutrient content in pulp of mango cv. Kesar.

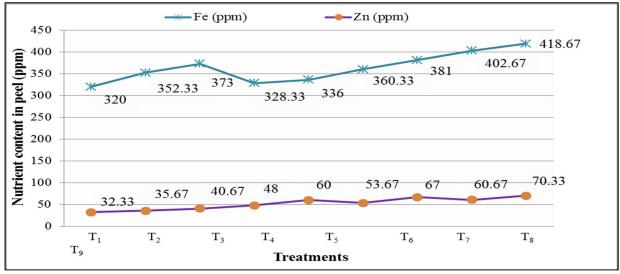


Figure 4: Effect of foliar spray of Zn and Fe on micro nutrient content in peel of mango cv. Kesar.

Kesar were recorded in the treatment combination of 0.50% FeSO₄ and 0.50% ZnSO₄ (T₉). In 2020, Li Li and his coworkers reported that the pulp may have been a greater sink for zinc than the other mango components. Fruits and leaves quickly absorb iron, increasing their capacity for concentration and uptake. Auxin and protein synthesis, seed development, and the activation of several enzymes are all impacted by zinc. All of these activities promote zinc absorption. The results of nutrient absorption closely support the findings of Sultana et al. (2018) in wheat, Chhetri et al.

(2017) in mandarin orange, and Dhaliwal *et al.* (2021) in lentil.

Conclusion

The results of the study indicate that the foliar application of 0.50% FeSO4 and 0.50% ZnSO4 nearly quadrupled the content of N, K, and P in pulp and peel, as well as Fe and Zn in pulp and peel. The quality of harvested crops pre- and postharvest is one advantage of foliar treatments. For even better results, this experiment can be improved by applying nanotechnology to the reducing malnutrition.

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leaves. However, this study will contribute to parents, friends and dear ones. I consider myself lucky to have worked under the guidance of

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Conflict of interest

The authors declare that they have no conflict of interest.

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